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The anticancer activity of chalcone compounds against human prostate carcinoma cell and human colon colorectal adenocarcinoma

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ABSTRACT

Cancer is the second most common cause of death in our country and also worldwide. In the cancer treatment are used many methods such as surgery, chemotherapy, radiotherapy and hormone therapy. Despite these methods, a decrease is not observed in cancer cases and in the mortality. In addition, cancer drugs cause serious side effects. Therefore, there is a need for new anticancer drugs. Chalcones have broad anticancer activity against various cancer cells. In this study, the anticancer activities of the chalcone compounds were investigated against PC3 (Human Prostate Carcinoma Cell) and HT29 (Human colon colorectal adenocarcinoma) cell lines. The anticancer activity tests were performed using the BrdU ELISA method. 5-Fluorouracil (5-FU) was used as the positive control. Compound 1-7 (except Compound 3) were determined having higher anticancer activity than 5-FU against PC3 cancer cells. IC50 values of the compounds 1-7 against PC3 cancer cells were 2.53±0.02 μM; 4.40±0.03 μM; 5.86±0.02 μM; 4.36±0.03 μM; 1.76±0.02 µM; 1.78±0.03 µM; 2.31±0.03 µM, respectively.

Keywords: HT29, PC3, Chalcone, 5-Fluorouracil, Fluor atom, anticancer activity.

Kalkon bileşiklerinin insan prostat karsinom hücresi ve kolon kolorektal adenokarsinomuna karşı antikanser aktivitesi

ÖZ

Kanser, ülkemizde ve dünya çapında en yaygın ikinci ölüm Kanser tedavisinde ameliyat, kemoterapi, nedenidir. radyoterapi ve hormon tedavisi gibi birçok yöntem kullanılmaktadır. Bu yöntemlere rağmen kanser vakalarında ve ölümlerde azalma gözlenmemektedir. Ayrıca kanser ilaçları ciddi yan etkilere neden olur. Bu nedenle yeni antikanser ilaçlara ihtiyaç vardır. Kalkonlar, çeşitli kanser hücrelerine karşı geniş bir antikanser aktivitesine sahiptir. Bu çalışmada kalkon türevlerinin antikanser aktiviteleri PC3 (insan prostat karsinom hücresi) ve HT29 (İnsan kolon kolorektal adenokarsinom) hücre hatlarına karşı araştırılmıştır. Antikanser aktivite testleri BrdU ELISA yöntemi kullanılarak yapıldı. Pozitif kontrol olarak 5-Florourasil (5-FU) kullanıldı. Bileşik 1-7 (Bileşik 3 hariç), PC3 kanser hücrelerine karşı 5-FU'dan daha yüksek antikanser aktiviteye sahip olduğu belirlendi. Bileşik 1-7'nin PC3 kanser hücrelerine karşı IC50 değerleri sırasıyla, 2.53±0.02 µM; 4.40±0.03 µM; 5.86±0.02 µM; 4.36±0.03 µM; 1.76±0.02 µM; 1.78±0.03 µM; 2.31±0.03 µM'dı.

Anahtar Kelimeler: HT29, PC3, Kalkon, 5-Flurourasil, Flor atomu, Antikanser aktivite.

1. INTRODUCTION

Cancer is the second most common cause of death in our country and also worldwide. One out of every 5 people

in the world gets cancer during their lifetime. 1 in every 8 men and 1 in every 11 women die of cancer.¹ The most common cancers seen in men are lung, prostate and colorectal cancers. In women, breast, lung, colorectal and

cervical cancers are seen the most common.¹ According to current estimates, 19.3 million new cases were diagnosed in 2020, and the cancer burden increased to 19.3 million new cases and 10.0 million deaths in 2020.¹ The top 5 most common types of cancer in our country in 2020 were lung cancer (17.6%), breast cancer (10.3%), colorectal (colon) cancer (9.1%), prostate cancer (8.3%) and thyroid cancer (5.9%). These 5 types of cancer accounted for more than 50% of all cancers.²

Worldwide, 14 million new cases of prostate cancer diagnosed and 375,304 prostate cancer (PCA) deaths occurred in 2020. The second most common cancer in men is PCA in 2020.¹ This type of cancer ranks fifth among cancer-related deaths. PCA is the most common cancer among men in more than half of the countries of the world. (105 out of 185); particularly in America, Northern and Western Europe, Australia/New Zealand and a large part of Sub-Saharan Africa.³ The natural course of this disease is quite heterogeneous and is still poorly understood. As a result of autopsy studies, prostate cancer is diagnosed in about one out of every three men over the age of 50. 80% of these tumours are 0.5 cm in size and low grade. In addition, the majority are considered to be clinically insignificant.⁴ Increasing incidences and mortality due to PCA and the failure of traditional chemo and radiotherapy of advanced invasive PCA show that essentially new approaches are needed to control this malignancy.^{5,6}

In 2020, there were more than 1.9 million new colorectal cancer diagnoses and 935,000 deaths. This situation shows that approximately one out of every 10 cases of cancer results in death. Generally, colorectal cancer ranks third in terms of the incidence and second in mortality.³ If diagnosed and treated at the early stages, colorectal cancer.⁷ Various systemic treatments such as surgery, chemotherapy, radiotherapy and hormone therapy are

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used for cancer treatment.^{8,9} Despite these treatment methods, a decrease is not observed neither in the number of patients with this disease nor in the mortality rate.⁹ In addition, cancer drugs cause toxicity in normal cells and tissues, causing serious side effects such as vomiting, nausea, hair loss, and resistance development.3,10-11 Therefore, there is a need for anticancer drugs that are and reliable, powerful, non-toxic, economical environment friendly. In a study we conducted in 2017, we synthesized the chalcone derivatives included in the present article and investigated the anticancer activities against HeLa and C6 cancer cells. As a result of our research, it was determined that these molecules showed higher anticancer effects than 5-Fluorouracil and Cisplatin, which were used as positive controls.¹² Thus, we examined the anticancer activities of these chalcone derivatives against PC3 (Human Prostate Carcinoma Cell) and HT29 (Homo sapiens colon colorectal adenocarcinoma) cell lines in the manuscript. The chalcone compounds in this study contain morpholine rings in the structure of cancer drugs such as Alectinib, Gefitinib and Canertinib (Figure 1). In addition, these chalcone compounds contain fluorinated organic structures such as 5-Fluorouracil, Floxuridin and Flutamid, which are also used in cancer treatment (Figure 1)

Chalcones consist of aryl rings linking to an α , β unsaturated ketone group. The easy synthesis of chalcones from various benzaldehyde and acetophenones enables them to be used in drug synthesis. Chalcones possess a number of biological effects such as antimalaria, antibacterial, antiinflammatory and anticancer.¹³

Fluorine atom provides two kinds of biological and chemical effects to the molecule. The organofluorine compounds have some features as shown in Table 1.¹⁴

Chemical effect	Biological effect	
Small size	Electronegativity of neighboring groups	
Lipophilic	Strong C-F bonds resistant to metabolic processes	
High electronegativity	th electronegativity Increases lipid solubility (bioavailability)	
Low reactivity	Synthesis of isosteric analogues of drugs Useful for studying biochemical processes	

Table 1. The chemical and	biological offects	of organofly oring com	nounda
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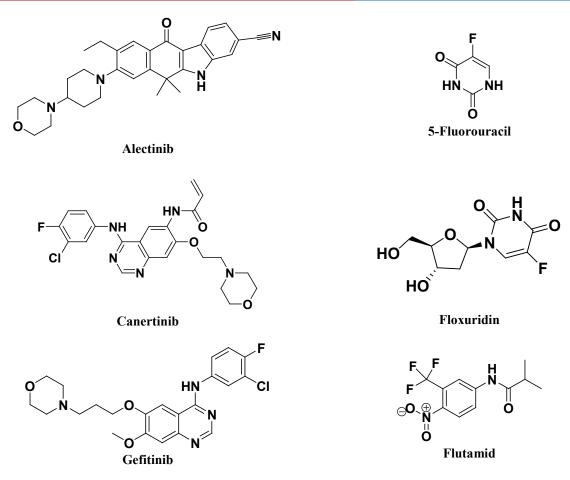


Figure 1. The anticancer drugs

2. MATERIALS AND METHODS

2.1. Preparation of compounds solution

The compounds and 5-Fluorouracil (5-FU) were dissolved in dimethyl sulfoxide (DMSO). The solution was diluted with Dulbecco's Modified Eagle Medium (DMEM). DMSO amount was below 0.1% in the solutions. 5-FU was used as the positive control.

2.2. Anticancer activity method

The tests were carried out on PC3 and HT29. The cells were purchased from the American Type Culture Collection (ATCC) with codes of HT-29 (ATCC® HTB38TM) and PC-3 (ATCC® CRL-1435TM). The cells were cultured in DMEM (10% of fetal bovine serum (FBS)- 2% of penicillin-streptomycin). DMEM was refreshed once in three days at 37 °C and 5% CO₂ in an incubator. 5-FU was used as the standard drug. The microplate wells were pipetted with 30.000 cells, then incubated for 30 minutes in CO₂ incubator. At the end of 30 h, the wells were added to the sample solutions on four concentrations. Then, the microplates were incubated

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overnight in same conditions. The tests were carried out in accordance with the BrdU ELISA kit protocol. Measurements were measured in ELISA reader at 450 nm. The activity was calculated as follows:

[1- (Atreatments /Avehicle control] x100.

 IC_{50} was calculated by ED50 plus v1.0, and the results were specified as means \pm SD of six values. The statistical analysis was made with SPSS 13.5 (p<0.01).

3. RESULTS AND DISCUSSION

Anticancer effects of all the samples and 5-fluorouracil (5-FU) were determined against HT29 cells at 100, 50, 25 and 5 μ M. The IC₅₀ values of all the samples were given at Table 2. The anticancer activities of all the samples were observed as showing dose-dependent increase in activity against HT29 cells (Figure 2). Compound 6 against HT29 cells were determined as having an effect fairly close to 5-FU. However, other compounds have better anticancer activities than 5-FU at high concentrations. The potency of inhibitions (at 100 μ M) against HT29 cells were: compound $3\sim$ compound 6> 5-FU> compound 2> compound $4\sim$ compound 1> compound 4> compound 5.

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Table 2. The IC₅₀ values (µM) of compounds 1-7 against HT29 and PC3 cancer cells

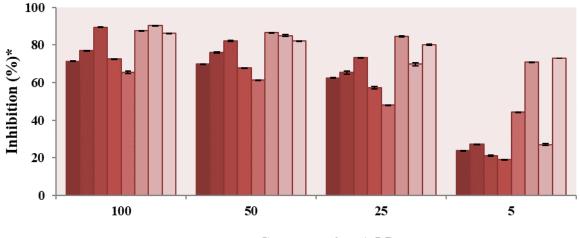
No	e IC ₅₀ values (μM) of compounds 1-7 against HT29 and PC3 ca Compound structure and name	IC 50	
		HT29	PC3
1	O F CF3 (E)-3-(2-fluoro-3-(trifluoromethyl)phenyl)-1-(4-morpholinophenyl) prop-2-en-1-one	6.26±0.03	2.53±0.02
2	O F CF CF CF CF CF F CF CF F CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF CF	5.50±0.02	4.40±0.03
3	O $FF_{3}C(E)-3-(2-fluoro-6-(trifluoromethyl)phenyl)-1-(4-morpholinophenyl) prop-2-en-1-one$	7.03±0.02	5.86±0.02
4	CF ₃ CF ₃ (<i>E</i>)-3-(4-fluoro-3-(trifluoromethyl)phenyl)-1-(4- morpholinophenyl) prop-2-en-1-one	7.89±0.05	4.36±0.03
5	O CF_3 (E)-3-(3,5-bis(trifluoromethyl)phenyl)-1-(4-morpholinophenyl) prop-2-en-1-one	3.38±0.01	1.76±0.02

Table 2. Continued

6	O OCF ₃ N (<i>E</i>)-1-(4-morpholinophenyl)-3-(2- (trifluoromethoxy)phenyl) prop-2-en-1-one	2.11±0.01	1.78±0.03
7	O O O O O O O O O O C F ₃ (<i>E</i>)-1-(4-morpholinophenyl)-3-(3- (trifluoromethoxy)phenyl) prop-2-en-1-one	5.52±0.02	2.31±0.03
5-FU		2.04±0.01	5.69±0.02

against HT29 cell line

■1 ■2 ■3 ■4 ■5 ■6 ■7 ■5-FU



Concentration (µM)

Figure 2. The anticancer activities of all the samples and 5-FU against HT29 cells. * tests were carried out as two experiments with triplicate (n=6; p<0.01).

Anticancer activities of all the samples and 5-FU were determined against PC3 cells at four concentrations (Figure 3). The IC_{50} values of compounds 1-7 are given at Table 2. The anticancer activities of compounds 1-7 were observed as showing dose-dependent increase in

activity against PC3 cells (Figure 3). All of the compounds (except Compound 3) were determined to have higher anticancer activities compared with 5-FU which was used as the standard compound against PC3 cells. Compound 3 against HT29 cells were determined

against PC3 cell line

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to have fairly close to the 5-FU. The potency of inhibitions (at 100 μ M) against PC3 cells were: compound 5~ compound 6> compound 7> compound 1> compound 4~ compound 2> 5-FU~ compound 3.

Compound 5 has the highest anticancer activity against PC3 cells (IC_{50:} $1.76\pm0.02 \mu$ M). Compound 5 contains two CF₃ groups. It contains the highest number of CF₃ groups compared to the groups found in other molecules.

The fluorine atom has slightly more lipophilic property than hydrogen. However, trifluoromethyl (CF₃) is much more lipophilic than methyl or chlorine. This property often has the most important role in improving pharmacological activity.¹⁵ This supports the high activity of our compound **5**.

1 2 3 4 5 6 7 5-**F**U

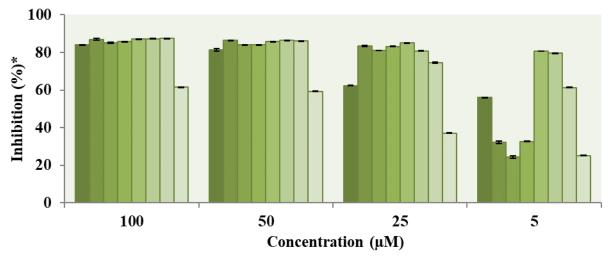


Figure 3. The anticancer activities of all the samples and 5-FU against PC3 cells. * tests were carried out as two experiments with triplicate (n=6; p<0.01).

The fluor and trifluoromethyl (electron-withdrawing groups) groups attached to the phenyl ring of chalcone in compound 1 are in position at 2-fluor and 3trifluoromethyl. The same groups are in position at 2fluor and 4- trifluoromethyl in compound 2. IC_{50} values for compound 1 and compound 2 against PC3 cells are 2.53±0.02 µM and 4.40±0.03, respectively. It was seen that the activity against PC3 cancer cells in the 2-3 position was higher than the one in the 2-4 position. F and CF₃ groups are in position at 2-fluor and 6trifluoromethyl in compound 3, and IC_{50} value of the compound is determined as 5.86±0.02 µM. When the fluorine and trifluoromethyl groups moved away from each other, a decrease in activity, such as compound 3, was detected. The anticancer activities of Compound 1-7 against PC3 were better than the ones against HT29.

The binding of a fluorine atom to a bioactive molecule causes minimal steric changes. Thus, it can facilitate the biomolecule's biological effects such as interaction with enzyme active sites, receptor recognition sites, and biological effects such as transport mechanisms. However, the addition of fluorine significantly changes the physicochemical properties of the bioactive molecule due to its great electronegativity. Therefore, these modified processes can affect biological activity.¹⁶ Bioisosteric derivatives of fluorine instead of the hydrogen atom are to further increase the activity of

biologically active molecules.^{14,17-19} Due to the chemical and biological properties of organofluorine compounds (Table 1), fluorine has become very important for predictions of new drugs and their development.^{15,20-22} Because of this, it affects the number of drugs that contain fluorine in the market today.²³ As in our study, fluorinated chalcone derivatives show anticancer activity.^{13,17,24,25}

4. CONCLUSIONS

Fluorine has chemical and biological properties. Thus it becomes very important for the planning and development of new drugs. Organofluorine structures affect the number of drugs that contain fluorine in the market today. In the study, fluorinated chalcone derivatives especially show higher anticancer activity against PC3 cancer cells compared to 5-FU.

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Conflict of interests

I declare that there is no a conflict of interest with any person, institute, company, etc.

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